Physical Principles in Biology Biology 3550 Spring 2024

Lecture 34

Protein Folding Thermodynamics and Mechanisms, and

Structure Prediction

Monday, 8 April 2024

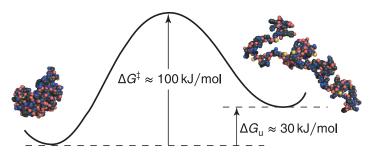
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### Announcements

- Problem Set 5:
  - Due Monday, 15 April at 11:59 PM
  - Submit pdf file on Gradescope
- Quiz 5:
  - Friday, 12 April
  - 25 min, second half of class
- Review Session:
  - 5:15 PM, Thursday, 11 April
  - HEB 2002
  - Come with questions!

# Protein Unfolding/Refolding: A Simplified Summary

Free energy profile for unfolding and refolding:



■ What determines the overall equilibrium between native and unfolded states?

### Conformational Entropy Change for Protein Unfolding

From the previous lecture:

$$\Delta S_{\rm conf} = k \ln 10^n$$

*n* is the number of amino acid residues. Assumes 1 conformation for the native state and 10 conformations for each residue in the unfolded state.

• On a molar basis for n = 100

$$\Delta S_{\mathsf{conf}} = R \ln 10^{100} = 8.314 \, \mathsf{J/(mol \cdot K)} \times \ln 10^{100}$$

$$= 2 \times 10^3 \, \mathsf{J/(mol \cdot K)}$$

■ Corresponding free energy change at 298 K:

$$-T\Delta S_{\rm conf} = -5.7 \times 10^5 \, \mathrm{J/mol} = -570 \, \mathrm{kJ/mol}$$

 Compare with the overall free energy change for unfolding, on the order of 30 kJ/mol

# Observed Thermodynamics for Protein Folding

For a typical small protein at room temperature (300 K):

Measured experimentally for unfolding:

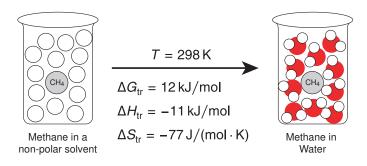
$$\Delta G_{\mathrm{u}} = 30\,\mathrm{kJ/mol}$$
  $\Delta H_{\mathrm{u}} = 100\,\mathrm{kJ/mol}$   $\Delta S_{\mathrm{u}} = 230\,\mathrm{J/(mol\cdot K)}$ 

Estimated change in conformational entropy:

$$\Delta S_{
m conf} = 2 imes 10^3 \, {
m J/(mol \cdot K)}$$
 $- T \Delta S_{
m conf} = -570 \, {
m kJ/mol}$ 

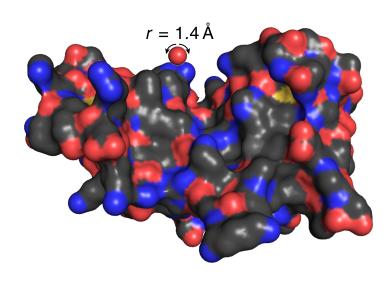
- What we need to explain:
  - Why is  $\Delta S_{\rm H} \ll \Delta S_{\rm conf}$ ?
  - Why is  $\Delta G_{\rm u} \gg -T\Delta S_{\rm conf}$ ?

### Thermodynamics of Transfer of a Non-polar Molecule to Water

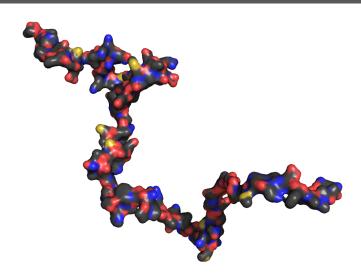


- $\Delta G_{\rm tr} = \Delta H_{\rm tr} T \Delta S_{\rm tr}$
- $\Delta G_{tr}$  is positive because  $\Delta S_{tr}$  is negative! (an "entropically driven" process).
- Water molecules become more ordered when a non-polar molecule is introduced.
- Non-polar groups buried in the interior of folded proteins become exposed to water on unfolding.

### Solvent-accessible Surface of a Small Protein: Ribonuclease A



# Solvent-accessible Surface of Unfolded Ribonuclease A (one representative conformation)



# Change in Accessible Surface Area for Unfolding for a Protein of About 100 Residues

	Folded (Å <sup>2</sup> )	Unfolded (Å <sup>2</sup> )	Difference (Å <sup>2</sup> )
Total	7,000	14,700	7,700
Non-polar	3,800	8,800	5,000
Polar	3,200	5,900	2,700

# Thermodynamic Consequence of Non-polar Surface Area Exposed Upon Unfolding (Hydrophobic Effect)

For  $5000 \, \text{Å}^2$  at  $300 \, \text{K}$ 

$$\Delta H_{\text{hyd}} = 35 \,\text{kJ/mol}$$

$$\Delta S_{\mathsf{hyd}} = -1,500\,\mathsf{J/(mol\cdot K)}$$

$$\Delta G_{\text{hvd}} = 480 \,\text{kJ/mol}$$

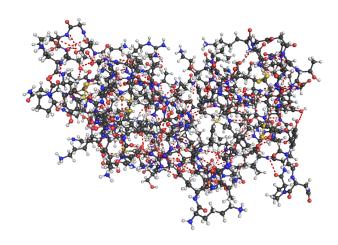
## Contributions to Protein Unfolding Thermodynamics

■ For protein of 100 amino-acid residues at 300 K:

	∆ <i>H</i> kJ/mol	Δ <i>S</i> J/(mol · K)	$\Delta G$ k $J/mol$
Conformational entropy	,	2,000	-570
Hydrophobic effect	35	-1,500	480
Other	65	-270	120
Overall, experimental	100	230	30

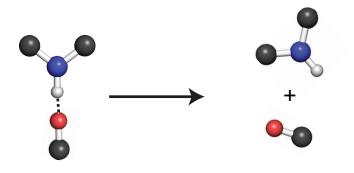
- Increase in conformational entropy is largely compensated for by decrease in water entropy associated with hydrophobic effect.
- What might "other" contributions to  $\Delta H$  be?
  - Breaking protein hydrogen bonds.
  - Exposure of polar surface area to water.

### Hydrogen Bonds in Folded Ribonuclease A



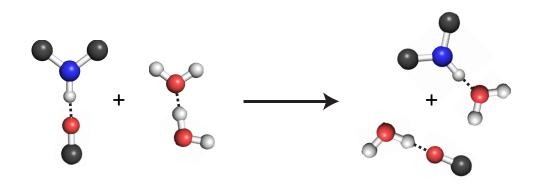
Red dashes indicate hydrogen bonds.

# Breaking a Hydrogen Bond in vacuo



lacksquare  $\Delta H pprox 50 \, kJ/mol$ 

# Breaking a Hydrogen Bond in Water



$$\Delta S = ?$$

$$\Delta G = ?$$

# Contributions to Protein Unfolding Thermodynamics

For protein of 100 amino-acid residues at 300 K:

	$\Delta H$	$\Delta S$	$\Delta G$
	kJ/mol	J/(mol·K)	kJ/mol
Conformational entropy		2,000	-570
Hydrophobic effect	35	-1,500	480
Other	65	-270	120
Overall	100	230	30

- Increase in conformational entropy is largely compensated for by decrease in water entropy associated with hydrophobic effect.
- Breaking hydrogen bonds likely represents much of the "other" contributions.

# Warning!

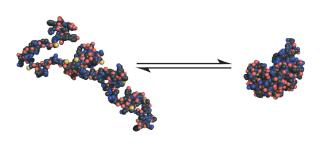


# **Direction Change**

HOW does the folded structure form?

### Clicker Question #1

For a protein of 100 amino-acid residues, how long would it take for the chain to randomly sample all of the possible conformations to find the native structure?



- A) Less than 1 second
- **B)**  $\approx 1$  minute
- C)  $\approx 1 \, \text{hour}$
- D)  $\approx 1 \, \mathrm{day}$
- E) More than 1 year

All answers count for now.

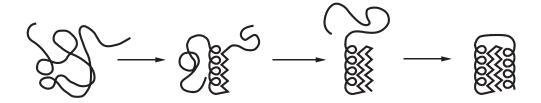
## The Levinthal paradox:

- Consider a polypeptide of 100 amino-acid residues.
- If conformations of individual residues are independent: 10<sup>100</sup> possible conformations.
- Assume that only 1 in 10<sup>10</sup> of these conformations is actually possible, because of steric conflicts, leaving 10<sup>90</sup> conformations.
- The fastest interconversions between conformations is on the order of  $10^{-13}$  s.

time = 
$$10^{90}$$
 conformations  $\times$   $10^{-13}$  s/conformation =  $10^{77}$  s =  $10^{77}$  s  $\div$   $3600$  s/hr  $\div$  24 h/day  $\div$  365 days/year  $\approx 10^{70}$  years

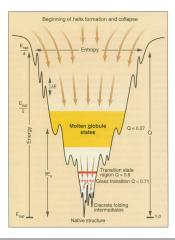
How does a polypeptide find it's folded conformation in seconds or minutes?

## Protein Folding as a Pathway



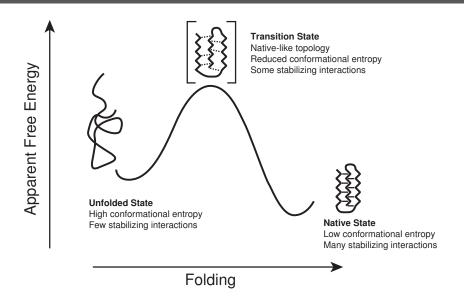
- Folding begins with a "nucleus" of local structure.
- Additional structure adds and increases stability.
- Rate-limiting step (transition state) might occur early or late in the pathway.

## Protein Folding as a Funnel

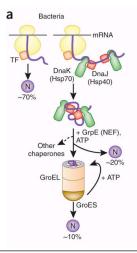


- Folding is viewed as a convergence of many possible starting conformations.
- Top of funnel represents unfolded state.
- Bottom of funnel represents native state.
- Width of funnel represents number of conformations (S<sub>conf</sub>).
- Distance from top to bottom represents number of stabilizing interactions.

### A Plausible Picture of the Transition State for Protein Folding

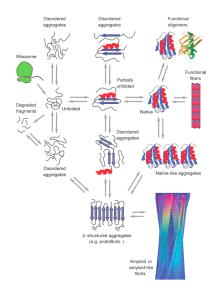


## Protein Folding in vivo



- Polypeptides are synthesized on ribosomes, starting with the N-terminus.
- Folding may begin on ribosomes.
- Molecular chaperones (Hsp70 and Hsp40) may limit folding before synthesis is complete.
- Other chaperones (GrpE and GroE) facilitate correct folding after synthesis.
- Chaperones have a largely negative role: preventing improper interactions.
- Some chaperones are ATP-driven machines that modify structures.

# Good Pathways and Bad



- Proteins are inherently "sticky".
- Many folded proteins assemble into functional oligomers and fibers.
- Unfolded or partially folded proteins are especially sticky.
- Unfolded or partially folded proteins tend to form aggregates or abnormal fibers (amyloids).
- Many diseases are associated with amyloid fibers. (Parkinson's disease, Alzheimer's disease, prion diseases).

Chiti, F. & Dobson, C. M. (2006). Protein misfolding, functional amyloid, and human disease. *Annu. Rev. Biochem.*, 75, 333–366. http://dx.doi.org/10.1146/annurev.biochem.75.101304.123901

# Some Approaches to Predicting Protein Structures

#### Hierarchical approach:

- Determine propensities of different amino acids to form  $\alpha$ -helices and  $\beta$ -strands.
- Use propensities to predict segments of polypeptide chain that will form  $\alpha$ -helices and  $\beta$ -strands.
- Assemble secondary-structure elements into overall fold.
- Doesn't really work!

#### ■ Template-based modeling:

- Identify a protein with a sequence very similar to the protein of interest, and with a known three-dimensional structure.
- Adjust the known structure to accommodate the sequence of the protein of interest.
- Works pretty well when the template structure is 50% or more identical to the unknown structure, but accuracy is limited.

# Some Approaches to Predicting Protein Structures

- Physics-based modeling:
  - Build a computer model of the polypeptide chain, in arbitrary conformation.
  - Apply mathematical functions that describe all of the forces acting on individual atoms.
  - Simulate process of sampling conformations to find those with minimum energies.
  - Provides information about the folding mechanism AND predicts structure!
  - Now feasible with very small proteins, but with high computational cost.