

Physical Principles in Biology

Biology 3550

Spring 2024

Lecture 17:

Diffusion from a Sharp Boundary

and

Kinetic Energy

Friday, 16 February 2024

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Announcements

■ Problem Set 3:

- Due 11:59 PM, Monday, 26 February.
- Download problems from Canvas.
- Upload work to Gradescope.
- Show your work!
- Please don't scrunch things up!

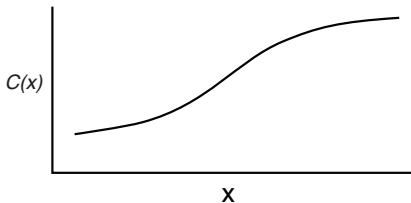
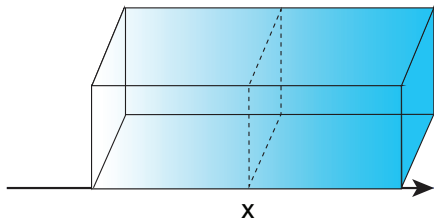
■ Quiz 3:

- Friday, 23 February
- 25 min, second half of class.

■ Review Session

- 5:15 PM, Thursday, 22 February
- HEB 2002
- Come with questions!

Fick's First and Second Laws of Diffusion



- First law:

$$J = -D \frac{dC}{dx}$$

- Flux, J , at position x is proportional to the concentration gradient at that position.

- Second law:

$$\frac{dC}{dt} = D \frac{d^2C}{dx^2}$$

- Rate of change in concentration at position x is proportional to the derivative of the concentration gradient.

Fick's Second Law of Diffusion

$$\frac{dC}{dt} = D \frac{d^2C}{dx^2}$$

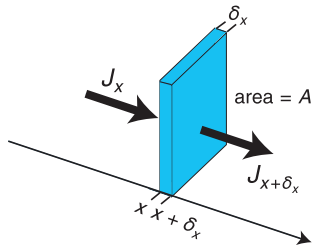
- A “second-order differential equation”.
- The solution to the equation is a function:

$$C = f(x, t)$$

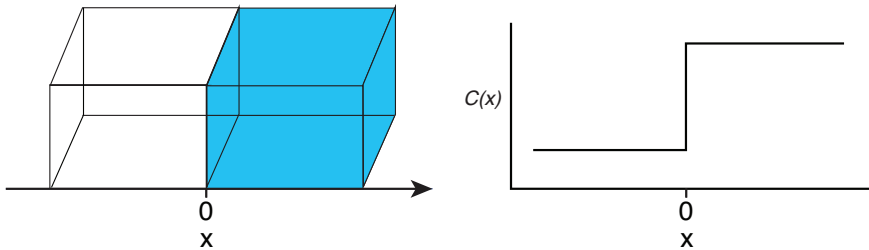
that satisfies the equation:

$$\frac{df(x, t)}{dt} = D \frac{d^2f(x, t)}{dx^2}$$

- The trick is to find $C = f(x, t)$.
- The solution depends on the shape of the volume and the initial concentrations, the *boundary conditions*.



Diffusion from a Sharp Boundary



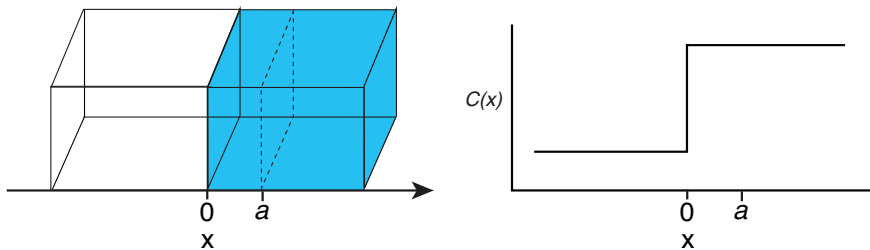
At $t = 0$

■ For $x < 0$: $C(x) = 0$, $\frac{dC}{dx} = 0$, $\frac{d^2C}{dx^2} = 0$

■ At $x = 0$: $\frac{dC}{dx} \rightarrow \infty$

■ For $x > 0$: $C(x) = 1$, $\frac{dC}{dx} = 0$, $\frac{d^2C}{dx^2} = 0$

Diffusion from a Sharp Boundary

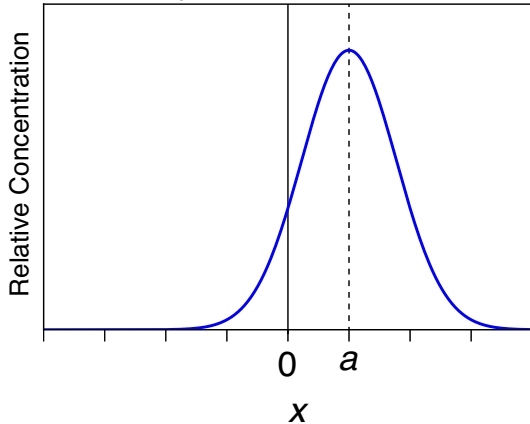


Consider molecules at a position $x = a > 0$:

- Molecules will begin to diffuse via a random walk.
- How will the molecules initially at position a be distributed after a time, t ?

Diffusion from a Sharp Boundary

Distribution of molecules originally at position $x = a$



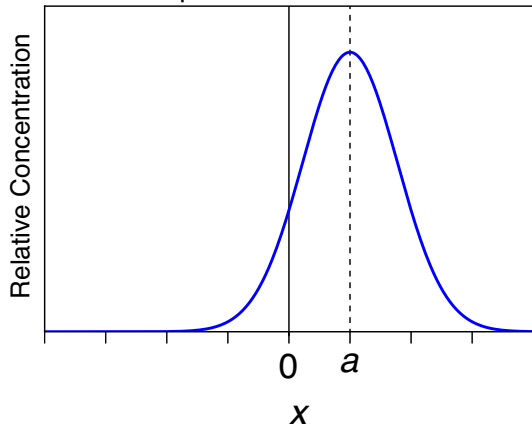
$$p(x) = \frac{1}{\sqrt{2\pi n \langle \delta_x^2 \rangle}} e^{-(x-a)^2 / (2n \langle \delta_x^2 \rangle)}$$

n = number of steps in random walk

$\langle \delta_x^2 \rangle$ = mean-square step distance along x -axis

Diffusion from a Sharp Boundary

Distribution of molecules originally at position $x = a$



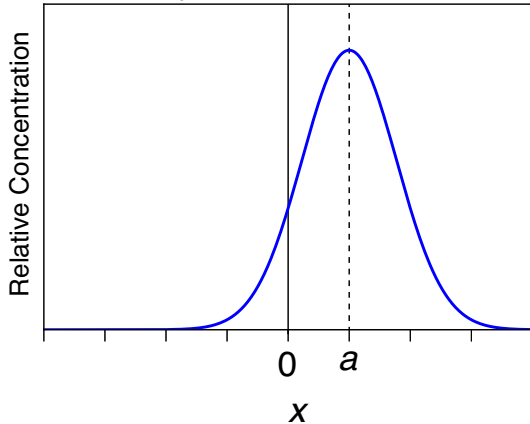
$$p(x) = \frac{1}{\sqrt{2\pi n \langle \delta_x^2 \rangle}} e^{-(x-a)^2 / (2n \langle \delta_x^2 \rangle)}$$

- Diffusion coefficient, $D = \frac{\delta_x^2}{2\tau}$
- τ = average time of each RW step
- After time, t , $n = t/\tau$

$$n \langle \delta_x^2 \rangle = \frac{t \langle \delta_x^2 \rangle}{\tau} = 2Dt$$

Diffusion from a Sharp Boundary

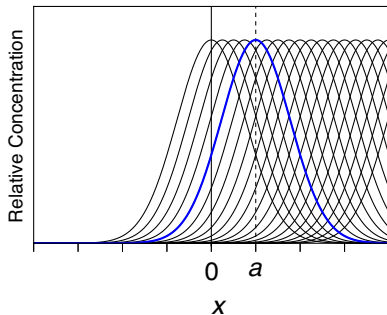
Distribution of molecules originally at
position $x = a$



$$p(x) = \frac{1}{\sqrt{4\pi Dt}} e^{-(x-a)^2/(4Dt)}$$

Diffusion from a Sharp Boundary

- Distribution of molecules from all starting points, $a \geq 0$.



- At position x , concentration is the sum of molecules that have diffused from $a \geq 0$

$$C(x, t) = \frac{1}{\sqrt{4\pi Dt}} \int_0^{\infty} e^{-(x-a)^2/(4Dt)} da$$

Does the “Solution” Satisfy Fick’s Second Law?

- Putative solution:

$$C(x, t) = \frac{1}{\sqrt{4\pi Dt}} \int_0^{\infty} e^{-(x-a)^2/(4Dt)} da$$

- Fick’s second law:

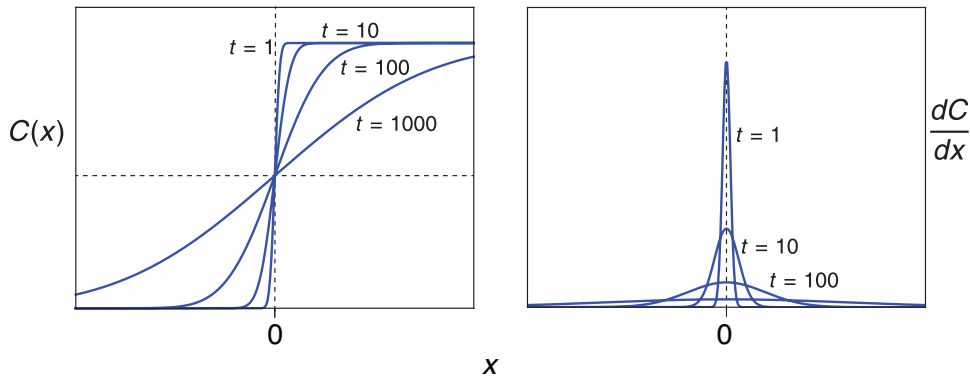
$$\frac{dC}{dt} = D \frac{d^2C}{dx^2}$$

- Need to evaluate $\frac{dC}{dt}$ and $\frac{d^2C}{dx^2}$ and see if they satisfy the equation.

They do!

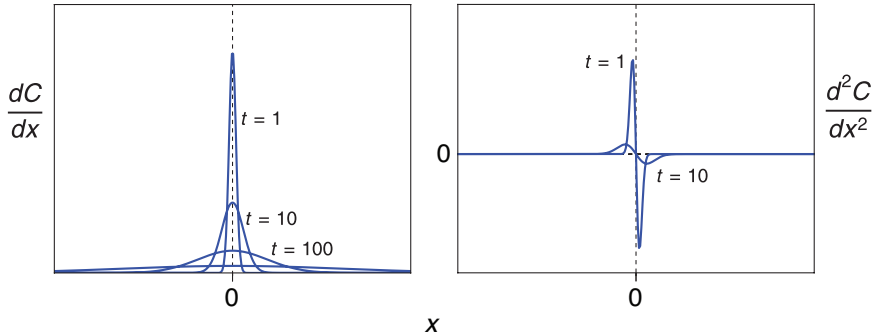
- $C(x, t)$ can’t be evaluated analytically, but it can be numerically.

Diffusion from a Sharp Boundary



- $C(x, t)$ is the integral of a Gaussian function with respect to x .
- The derivative of $C(x, t)$ with respect to x is a Gaussian function!

Diffusion from a Sharp Boundary



- Concentration increases most rapidly where the second derivative is most positive.
- Concentration decreases most rapidly where the second derivative is most negative.
- Concentration does not change at $x = 0$, where the flux, J , is greatest!

Estimating D from Diffusion from a Sharp Boundary

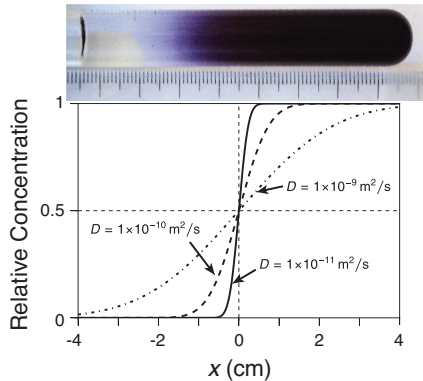
$$C(x, t) = \int_0^{\infty} \frac{1}{\sqrt{4\pi Dt}} e^{-(x-a)^2/(4Dt)} da$$

$$t = 48 \text{ hr} = 1.7 \times 10^5 \text{ s}$$

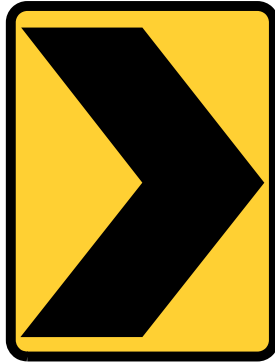
$$D \approx 2 \times 10^{-10} \text{ m}^2/\text{s}$$

$$D = \frac{\delta_x^2}{2\tau}$$

- What are the average length (δ_x) and duration (τ) of the random walk steps?
- We could answer these questions if we knew the velocity of the molecules, δ_x/τ .



Warning!



Direction Change

Molecular Motion and Kinetic Energy

Molecular Motion and Kinetic Energy

■ What is energy?

Capacity to do work.

■ What is work?

Mechanical work: The application of force over distance:

$$w = \int_a^b F dx$$

■ The units of work and energy.

- Force: Units defined by Newton's second law: $F = \text{mass} \times \text{acceleration}$

SI unit of mass: Kg

Acceleration: change in velocity (m/s) with time. SI units: m/s^2

SI units of Force: $\text{Kg} \cdot \text{m/s}^2$

$$1 \text{ N} = 1 \text{ Kg} \cdot \text{m/s}^2$$

- Work or energy: $\text{Kg} \cdot \text{m}^2/\text{s}^2$

$$1 \text{ J} = 1 \text{ N} \cdot \text{m} = 1 \text{ Kg} \cdot \text{m}^2/\text{s}^2$$

Kinetic Energy

- A object of mass, m , moving with velocity, v , in the x -direction has kinetic energy in that direction of:

$$E_{k,x} = mv_x^2/2$$

Check the units: $\text{Kg} \times (\text{m/s})^2 = \text{Kg} \cdot \text{m}^2/\text{s}^2$ It's OK!

- What does this mean?
 - The energy required to accelerate the mass, m , from rest to velocity, v_x .
 - Also the energy released during the deceleration of the mass from velocity, v_x , to rest.
 - Kinetic energy does not depend on the rate of acceleration or deceleration, only the final velocity.
 - But, amount of wasted energy likely does depend on rate of acceleration!
 - $E_{k,x}$ is proportional to v_x^2 . What are the implications?

Clicker Question #1

What is the kinetic energy of a baseball ($m = 145 \text{ g}$) with a velocity of 40 m/s ($\approx 90 \text{ miles/h}$)?

- A) 3 J
- B) $50 \text{ kg} \cdot \text{m}^2/\text{s}^2$
- C) 100 Nm
- D) 200 J
- E) 200 N

$$E_k = \frac{1}{2}mv^2 = \frac{1}{2}0.145 \text{ Kg}(40 \text{ m/s})^2 = 120 \text{ Kg} \cdot \text{m}^2\text{s}^{-2} = 120 \text{ Nm} = 120 \text{ J}$$